

SODIUM CARBONITE

Product Overview:

• Product Name: Sodium Carbonate

• **CAS Number**: 497-19-8

• IUPAC Name: Disodium carbonate

Molecular Formula: Na₂CO₃

• **Synonyms**: Soda Ash, Disodium carbonate, Soda crystals, Sodium carbonate anhydrous, Washing soda, and more.

Chemical and Physical Properties:

• Molecular Weight: 105.99 g/mol

Density: 2.54 g/cm³
Melting Point: 856 °C

• **Boiling Point**: Decomposes on heating by CO₂ loss.

• **Solubility**: Freely soluble in water, insoluble in ethanol.

• Appearance: White hygroscopic powder, colorless crystals, or grayish-white lumps.

• Refractive Index: 1.535

• **pH**: Aqueous solutions are strongly alkaline.

✓ Safety and Hazards:

- **Primary Hazard**: Irritant (GHS07 symbol)
- **Safety Precautions**: Avoid contact with eyes and skin. Wear suitable protective clothing, gloves, and eye/face protection.
- **Handling**: Store in a dry place, away from acids and moisture.

Regulatory Information:

- **FDA**: Approved as an **active ingredient** for veterinary and human use.
- **EU Pesticides Database**: Not approved for pesticide use under Regulation (EC) No 1107/2009.
- **EINECS Number**: 207-838-8 (for regulatory and classification purposes in Europe).

✓ Pharmacological Data:



- **Therapeutic Uses**: Used topically for dermatitides and as a mouthwash or vaginal douche. Also used for emergency veterinary applications as an emetic.
- Mechanism of Action: Acts as an alkalizing agent, neutralizing stomach acid.

✓ Structural Information:

• 2D Structure:

3D Conformer:



✓ Spectral Data (Optional):

• NMR Spectra: Available upon request.

• IR Spectra: Available upon request.

• Raman Spectra: Available upon request.

✓ Vendors and Supply Chain:

• **Suppliers**: Chemical vendors, curators, and government organizations supply Sodium Carbonate.

• Purchase Links: Available via external links to vendors and chemical suppliers.

✓ Related Chemicals:

• **Related Compounds**: Sodium Carbonate Monohydrate (CID-2735133), Sodium Carbonate Decahydrate (CID-151402), and more.

Price and Availability:

• **Bulk Pricing**: Available on request.

• **Packaging**: Typically available in 25 kg, 50 kg, or custom packaging sizes.

✓ Applications:

1. Glass Industry:

- Sodium Carbonate is a key component in the production of glass, particularly soda-lime glass. It is used as a fluxing agent in the manufacturing of glass products such as windows, containers, and fiberglass.
- **Material Interaction**: Combines with **silica** (SiO₂) to lower the melting point of the mixture, facilitating the creation of glass at lower temperatures.

2. Detergents and Cleaning Products:



- Sodium carbonate, commonly referred to as **washing soda**, is used in **laundry detergents** and household **cleaning products**. It helps soften water and aids in the removal of grease, oil, and dirt.
- **Material Interaction**: Acts as a **water softener**, preventing calcium and magnesium ions from interfering with the cleaning action.

3. Water Treatment:

- Used in water softening and treatment processes to reduce water hardness. It reacts with calcium and magnesium ions in the water to form insoluble salts.
- Material Interaction: Forms calcium carbonate (CaCO₃) and magnesium carbonate (MgCO₃), which are precipitated out of the water, making it softer.

4. Textile Industry:

- In **textile processing**, **Sodium Carbonate** is used to remove **dyes** and to adjust the pH in **bleaching** and **washing** processes.
- **Material Interaction**: Facilitates the breakdown of organic impurities during the textile **dyeing process** and aids in **stabilizing pH** levels in various textile treatments.

5. Petroleum Refining:

- Sodium carbonate is used in **petroleum refining** to treat crude oil and remove sulfur compounds (desulfurization) and other impurities.
- **Material Interaction**: Neutralizes acidic impurities, improving the quality of refined petroleum products.

6. Chemical Manufacturing:

- It serves as an important **intermediate compound** in the production of various **chemicals**, such as **sodium bicarbonate** (baking soda), **sodium silicate**, **sodium phosphate**, and others.
- **Material Interaction**: Reacts with **carbon dioxide** to produce sodium bicarbonate and combines with silica to form sodium silicate, both of which are used in other chemical processes.

7. Paper Industry:

- **Sodium Carbonate** is used in the **kraft process** for wood pulp production in the paper industry. It helps in breaking down the lignin in wood, making the cellulose fibers easier to extract.
- **Material Interaction**: Acts as a **chemical pulping agent**, allowing for the extraction of cellulose fibers needed for paper.

8. Food Industry:

- **Sodium Carbonate** is used in the food industry as a **leavening agent**, **acidity regulator**, and **buffering agent**.
- **Materials Interaction**: Reacts with **acids** to release carbon dioxide gas, which aids in **leavening** baked goods like crackers and cookies.

9. Pharmaceuticals:

• Used in the formulation of **oral antacid products**, where it neutralizes stomach acid.



• Material Interaction: Reacts with gastric hydrochloric acid (HCl) to produce carbon dioxide (CO₂), which relieves indigestion and heartburn.

10. Agriculture:

- Sodium Carbonate is used in agriculture for soil pH adjustment and as a fungicide.
- **Material Interaction**: Increases soil **alkalinity**, making it suitable for crops that prefer alkaline conditions.

11. Mining:

- **Sodium Carbonate** is involved in the extraction of **metals** and minerals from ores, particularly in processes like **uranium and lithium extraction**.
- **Material Interaction**: Acts as a flux in smelting processes to reduce the melting point of ores.

12. Cosmetics and Personal Care:

- **Sodium Carbonate** is used as a **pH regulator** in cosmetics and personal care products, including **shampoos**, **soaps**, and **bath products**.
- Material Interaction: Ensures the products maintain the correct pH balance for skin compatibility.

13. Ceramics:

- Used in **ceramics** and **porcelain production** as a **fluxing agent** to lower the melting point of silica, aiding in the creation of ceramics and porcelain.
- **Material Interaction**: Reduces the **melting temperature** of the mixture of clay and silica, allowing it to form the desired shape at lower temperatures.

14. Laboratory Reagents:

- Used in **chemical laboratories** for **titration** processes, **buffer solutions**, and as a **reagent** in various analytical techniques.
- **Material Interaction**: Used to create **buffer solutions** that maintain a stable pH during chemical reactions.

15. Battery Manufacturing:

- Sodium Carbonate is used in the production of sodium-ion batteries as a source of sodium in some battery technologies.
- **Material Interaction**: Provides the **sodium ions** necessary for the battery's electrochemical reactions.

16. Fire Extinguishers:

- Sodium carbonate is used in the manufacturing of dry chemical fire extinguishers.
- Material Interaction: Acts as a fire suppressant, particularly for grease fires.